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## COLSID AP

TECHNICAL DATA  
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## COLSID AP

*PROCESS FOR BRIGHT CHLORIDE ZINC PLATING BATHS CONTAINING AMMONIUM CHLORIDE AND POTASSIUM CHLORIDE*

COLSID AP	produces a brilliant, level, ductile zinc deposit from acid chloride zinc plating electrolytes specially designed for mixed NH <sub>4</sub> Cl/KCl baths.
COLSID AP	baths do not require boric acid, eliminating an extra expense, time consuming analysis, and crystallization on anodes in cold weather.
COLSID AP	baths can operate at extremely high temperatures of 125+° F (52+° C).
COLSID AP	contains a special ductilizer compound which ensures extremely ductile, bright deposits.
COLSID AP	operates with little or no foam and can be used with air agitation and evaporative recovery systems.
COLSID AP	is extremely tolerant to organic contaminants such as oils/cleaner surfactants and inorganic contaminants such as iron/copper.
COLSID AP	deposits readily accepts conventional and high corrosion trivalent Passivate technologies.

## OPERATING PARAMETERS

### RACK AND BARREL BATHS

Zinc Metal:	2.0 - 5.0 oz/gal (15 - 37.5 g/L)
Chloride Ion:	16 - 22 oz/gal (120 - 165 g/L)
COLSID AP CARRIER:	3 - 5% by volume
pH (Electrometric):	5.0 - 6.0
Temperature:	75 - 125° F (24 - 52° C)

## SOLUTION MAKEUP

	<u>100 LITERS</u>	<u>100 GALLONS</u>
Zinc Chloride:	4.0 kg	33 lbs.
Potassium Chloride:	19.0 kg	158 lbs.
Ammonium Chloride:	3.5 kg	29 lbs.
COLSID AP CARRIER:	4.0 L	4 gal
COLSID AP BRIGHTENER:	100 mL	385 mL
COLSID AP STARTER:	0.5 L	0.5 gal

The bath is prepared by dissolving zinc chloride, potassium chloride, and ammonium chloride in hot water. The volume of the hot water should be roughly two-thirds the final volume of the plating solution. After all the salts have been thoroughly dissolved, **dilute the required amount of COLSID AP CARRIER with at least equal parts of water**; add to the bath and mix until dispersed. Follow the same procedure with the COLSID AP STARTER and the COLSID AP BRIGHTENER. Dilute the bath to the final volume and mix well.

Make-up of the bath as recommended above will yield the following bath chemistry:

Zinc Metal:	2.5 oz/gal (18.8 g/L)
Chloride Ion:	18.0 oz/gal (135 g/L)
Ratio KCl:NH <sub>4</sub> Cl*:	5:1
pH (Electrometric):	5.6

\*The ratio between potassium chloride and ammonium chloride is generally maintained at approximately 5:1, i.e., for every 100 pounds of chloride ion required, as determined by analysis, add 175 pounds potassium chloride and 34 pounds ammonium chloride.

## EQUIPMENT

Filtration	Continuous filtration through polypropylene filter tubes of approximately 15 microns is recommended for routine operation. When carbon treatment or other bath purification is necessary, 5 - 10-micron filter tubes should be substituted.
Equipment	All plating tanks, racks, carriers, etc., which come into contact with COLSID AP solutions should be plastisol, polyethylene, hard rubber, or similarly coated to provide adequate protection from corrosion.
Agitation	Unlike many competitive processes, COLSID AP does not foam excessively and both mechanical and air agitation can be used.
Ventilation	The spray from COLSID AP solutions (not fumes) is inherently corrosive. The use of fiberglass, PVC, or polyethylene ventilation equipment and exhaust fans is recommended to prolong equipment life.

Cooling      Cooling coils made from Teflon are optimum, but titanium coils may be used as long as they are insulated from the electrical circuit. Lead or steel coils are not suitable.

## MAINTENANCE ADDITIONS

COLSID AP BRIGHTENER is maintained in the plating bath at the rate of one gallon per 15,000 - 20,000 ampere-hours. Additions should be made every 2 - 4 hours of operation or continuously through the use of an addition agent pump.

COLSID AP-O BRIGHTENER is a highly concentrated solvent-containing brightener added to the plating bath at the rate of one gallon per 25,000 - 30,000 ampere-hours. Additions should be made every 2 - 4 hours of operation or continuously through the use of an addition agent pump.

COLSID AP STARTER is added only at the time of initial make up.

COLSID AP CARRIER must be replaced in the plating bath as it is lost by drag-out. COLSID AP CARRIER can also be lost due to frequent treatment for iron. The most efficient and effective way to replace COLSID AP CARRIER is to add it along with COLSID AP BRIGHTENER additions, as follows:

LOW DRAGOUT/LOW IRON CONTAMINATION (Most rack baths):

add 1-gallon AP CARRIER per 4 - 5 gallons AP BRIGHTENER  
(add 1-liter AP CARRIER per 4 - 5 liters AP BRIGHTENER)

MEDIUM DRAGOUT (Most Barrel Baths):

add 1-gallon AP CARRIER per 1-gallon AP BRIGHTENER  
(add 1-liter AP CARRIER per 1-liter AP BRIGHTENER)

HIGH DRAGOUT/HIGH IRON CONTAMINATION (Some barrel baths):

add 2 gallons AP CARRIER per 1-gallon AP BRIGHTENER  
(add 2 liters AP CARRIER per 1-liter AP BRIGHTENER)

Maintenance of COLSID AP additives should be checked by periodic Hull cell evaluations.

## TYPICAL CYCLE

### PRE-PLATE TREATMENT

The following cleaning and pickling cycle are recommended when using the COLSID AP process:

1. HOT ALKALINE SOAK CLEAN.
2. HOT ALKALINE ELECTRO-CLEAN (ANODIC CURRENT / ANODIC CLEANING).
3. RINSE.
4. 30 - 50% MURIATIC ACID PICKLE WITH 1 - 2% PICKLE PAL® OR PICKLE PAL® PLUS.
5. RINSE.
6. PLATE.

### POST-PLATE TREATMENT

COLSID AP deposits are whiter and brighter than many competitive systems and provide surfaces that are highly receptive to most conventional hexavalent and high corrosion trivalent technologies.

## ANALYTICAL PROCEDURE

### ANALYSIS FOR ZINC METAL

1. Pipette 5 mL bath sample into a 250 mL Erlenmeyer flask and add 100 mL distilled water.
2. Add 10 mL ammonium hydroxide buffer solution.
3. Add approximately 0.2 grams Eriochrome Black T Indicator Mix.
4. Add 10 mL 8% formaldehyde solution.
5. Titrate immediately with Standard EDTA Solution 0.0575 M to a blue endpoint.

FACTOR: (mL Standard EDTA Solution 0.0575 M) x 0.10 = oz/gal zinc metal. (oz/gal x 7.5 = g/L)

### ANALYSIS FOR TOTAL CHLORIDE

1. Pipette 10 mL bath sample into a 250 mL volumetric flask. Dilute to 250 mL with distilled water and mix well.
2. Pipette 10 mL of above dilute solution into a 500 mL Erlenmeyer flask and add 100 mL distilled water.
3. Add 5 mL Sodium Passivate Indicator.
4. Titrate with Standard Silver Nitrate Solution 0.153 N to a reddish-brown endpoint. (The first permanent brown color is the endpoint.)

FACTOR: (mL Standard Silver Nitrate Solution 0.153 N) x 1.82 = oz/gal Chloride. (oz/gal x 7.5 = g/L)

### PREPARATION OF ANALYTICAL REAGENTS

EDTA 0.0575 M	Dissolve 21.6 grams C.P. Di-sodium EDTA salt in distilled water; dilute to exactly one liter.
8% Formaldehyde Solution	Dilute 200 mL 40% C.P. Formaldehyde to one liter with distilled water.
Silver Nitrate 0.153 N	Add 6 mL Nitric Acid to 26.0 grams C.P. Silver Nitrate. Dissolve in distilled water, dilute to exactly one liter.
Eriochrome Black "T" Indicator Mix	Grind together 1-part indicator and 100 parts sugar.
Sodium Passivate Indicator	Dissolve 10.0 grams Sodium Passivate in 100 mL distilled water.

## HELPFUL HINTS

### pH

The pH of the plating bath should be maintained within operating limits by the addition of dilute hydrochloric acid. Care should be taken during pH adjustment, as the pH changes rapidly with small additions of acid. Hydrochloric acid should be diluted with equal parts water prior to adding to the plating bath to avoid localized precipitation of addition agents. The pH should be checked using a meter, not pH paper. A high pH, over 6.0, causes low current density dullness. Low pH conditions are not normally encountered and occur only if excessive amounts of hydrochloric acid have been added during pH adjustment. If this occurs, diluted ammonium hydroxide or potassium hydroxide may be added to raise the pH to the proper operating level. Ammonium hydroxide is preferred because it is instantly soluble in the bath.

**Ammonium Chloride/  
Potassium Chloride**

Ammonium Chloride and Potassium Chloride are the only chemical additions required in the plating bath during normal operation. They should be added on a regular basis based on periodic analysis for chloride.

The following equivalents should be noted when maintaining the chloride content of the bath:

Ammonium Chloride  $\text{NH}_4\text{Cl}$ , contains 66% chloride

Potassium Chloride,  $\text{KCl}$ , contains 48% chloride

Zinc Chloride,  $\text{ZnCl}_2$ , contains 52% chloride

Therefore, to raise chloride content 1.0 oz/gal or 7.5 g/L requires:

0.34 oz/gal or 2.5 g/L Ammonium Chloride and 1.75 oz/gal 13 g/L Potassium Chloride

**Zinc Metal**

Zinc Metal is normally maintained by anodic dissolution during electrolysis. High grade zinc slab anodes with a minimum of 99.99% purity are recommended as an economical anode source. Anodes may be drilled and tapped or used in titanium anode baskets. Dissolution of the zinc anode does not occur during idle periods and anodes do not have to be removed from the bath during shutdowns. It is recommended to maintain as much anode area as possible to promote good current distribution. Anode baskets should be kept full. Acid resistant anode bags of cotton, dyneel, or polypropylene are recommended for rack operation to reduce anode-caused roughness.

**Operating Temperature**

COLSID AP baths have a very wide range of bright operating temperatures, from approximately 70 - 120+° F (22 - 48+° C). The optimum temperature range for best brightness at minimum brightener consumption is 80 - 85° F (25 - 28° C) and the bath should be maintained at this temperature whenever economically feasible by cooling coils or refrigeration systems.

**HANDLING & STORAGE**

Columbia Chemical recommends referring to the specific product Safety Data Sheets for safety, handling, and storage precautions.

**NON-WARRANTY**

The data contained in this bulletin is believed by Columbia Chemical Corp. to be accurate, true, and complete. Since, however, final methods of use of this product are in the hands of the customer and beyond our control, we cannot guarantee that the customer will obtain the results described in this bulletin, nor can we assume responsibility of the use of this product by the customer in any process which may infringe the patents of third parties.